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## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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(54) Title: MONITORING METABOLIC CONTROL IN DIABETIC PATIENTS BY MEASURING GLYCOSYLATED AMINO ACID AND PEPTIDES IN URINE

(57) Abstract

Metabolic control in diabetic patients is monitored by measuring the amount of non-enzymatic glycosylated amino acids and peptides in the urine of the patient. The method comprises treating the urine with a boronic acid to form a complex with the non-enzymatic glycosylated amino acids, peptides or mixtures thereof. The complex is separated and quantitatively analyzed for amino acids, peptides or mixtures thereof. A test kit for the measurement is also disclosed.

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Description

MONITORING METABOLIC CONTROL IN  
DIABETIC PATIENTS BY MEASURING GLYCOSYLATED  
AMINO ACID AND PEPTIDES IN URINE

5      Technical Field

This invention relates to a method for separating and quantitating the degradation products of non-enzymatic glycosylated proteins.

Background Art

10       Glucose has been found to covalently modify a variety of proteins non-enzymatically in vivo, including hemoglobin, crystallins of lens protein, serum albumin and erythrocyte cell membrane. (Koenig et al, "Correlation Of Glucose Regulation And Hemoglobin A<sub>1C</sub> In Diabetes Mellitus", N. Engl. J. Med., 295, 417-420 (1976); Stevens et al, "Diabetic Cataract Formation: Potential Role Of Glycosylation Of Lens Crystallins", Pro. Natl. Acad. Sci. U.S.A., 75, 2918-2922 (1979); Day et al, Non-enzymatically Glucosylated Albumin", J. Biol. Chem. 254, 595-597 (1979); and Miller et al, "Non-enzymatic Glycosylation Of Erythrocyte Membrane Proteins", J. Clin. Invest., 65, 896-901 (1980)). This non-enzymatic glycosylation reaction most probably occurs with other proteins of the body as well, particularly in the insulin-independent tissues of diabetic patients (Cerami et al, "Role Of Non-enzymatic Glycosylation In The Development Of The sequelae Of Diabetes Mellitus", Metabolism, 28, 431-437 (1979)). The concentration of glycosylated hemoglobin has been shown to reflect mean blood glucose concentration during the preceding several weeks, and has thus provided an



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important new measurement for monitoring metabolic control in diabetic patients. The utility of this indicator in out-patient management of diabetes is somewhat limited, however, by the impracticality of  
5 making this measurement at frequent intervals at home, i.e., the taking of blood samples on a regular basis.

Additionally various techniques have been developed in the prior art for testing for proteinaceous materials, glucose and blood sugar conditions.

10 Atkinson et al., U. S. Patent 3,438,737, discloses a test for proteinaceous materials in various fluids, including urine, using a chromogenic indicator taking advantage of the so-called "protein error" phenomena. However, there is no indication that the proteinaceous  
15 materials are glycosylated amino acids and proteins, let alone that the test is more specific than "proteins". Hack et al., U. S. Patent 3,443,903, discloses, very generally, test papers for the detection of glucose, proteins and the like in biological fluids such as urine. There is no suggestion of a test for glycosylated amino acids and peptides. Acuff, U. S. Patents  
20 4,168,147; 4,142,856; 4,142,857; and 4,142,858 discloses methods for determining the blood sugar condition of diabetes patients by measuring the level of a  
25 specific hemoglobin species, Hb- $\text{A}_{1a-c}$ , in whole blood. There is no suggestion of testing urine.

A need therefore continues to exist for a method for monitoring metabolic control in diabetic patients which is practical for home use in the management of  
30 out-patients.

Disclosure of Invention

Accordingly, one object of the invention is to provide a method for monitoring metabolic control in diabetic patients which is practical for out-patient



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management.

Another object of this invention is to provide a method for the measurement of the degradation products of non-enzymatically glycosylated proteins.

5 Another object of this invention is to provide a test kit for the measurement of the degradation products of non-enzymatically glycosylated proteins in urine.

10 Briefly, these objects and other objects of the invention as hereinafter will become more readily apparent can be attained by providing a method for separating and quantitating the degradation products of non-enzymatic glycosylated proteins.

#### Brief Description of the Drawings

15 A more complete appreciation of the invention and many of the attendant advantages thereof will be readily obtained as the same become better understood by reference to the following detailed description when considered in connection with the accompanying  
20 drawings, wherein:

FIGURE 1 is a plot showing the separation of  $^{14}\text{C}$ -glycosylated lysine from a mixture of  $^3\text{H}$ -lysine and  $^{14}\text{C}$ -glycosylated lysine.

25 FIGURE 2 is a plot showing the elution of a large peak of ninhydrin-positive material in the void volume for urine samples using an alkaline buffer.

FIGURE 3 is a plot showing the elution of a single peak of ninhydrin-positive material, after alkaline washing, for urine samples eluted with HCl.

30 FIGURE 4 is a comparison of the levels of degradation products of non-enzymatic glycosylated proteins in urine from diabetic and normal patients.



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FIGURE 5 is a comparison of the levels of non-enzymatically glycosylated amino acids found in the peripheral nerve of normal and diabetic rats.

FIGURE 6 is a representation of a test strip  
5 wherein (a) is a front view and (b) is a side view.

FIGURE 7 is a representation of a test stick where  
(a) is a side view and (b) is an end view.

FIGURE 8 is a representation of a dye packet.

Best Mode For Carrying Out The Invention

10 This invention relates to the measurement of glycosylated amino acids and/or peptides in urine, by, separating and quantitating these compounds. This assay reflects changes in metabolic control, since it measures degradation products of glycosylated proteins.  
15 As previously noted, the concentration of non-enzymatically glycosylated proteins in blood has been shown to reflect mean blood glucose concentration during the preceding several weeks. Also, as previously noted, glycosylation of numerous other proteins  
20 occurs in the body. It has now been found that the metabolic degradation products of these proteins, i.e., non-enzymatically glycosylated amino acids and peptides, also reflect the mean blood glucose concentration. This in itself, is surprising since enzymatically glycosylated proteins which are broken down  
25 by the body do not appear as glycosylated peptides or amino acids in the urine or appear in such small quantities and relatively constant concentrations as to have no affect on the technique of this invention. It  
30 is the discovery that the degradation products of non-enzymatically glycosylated proteins appear as glycosylated amino acids and/or peptides in the urine which forms the basis of this invention.



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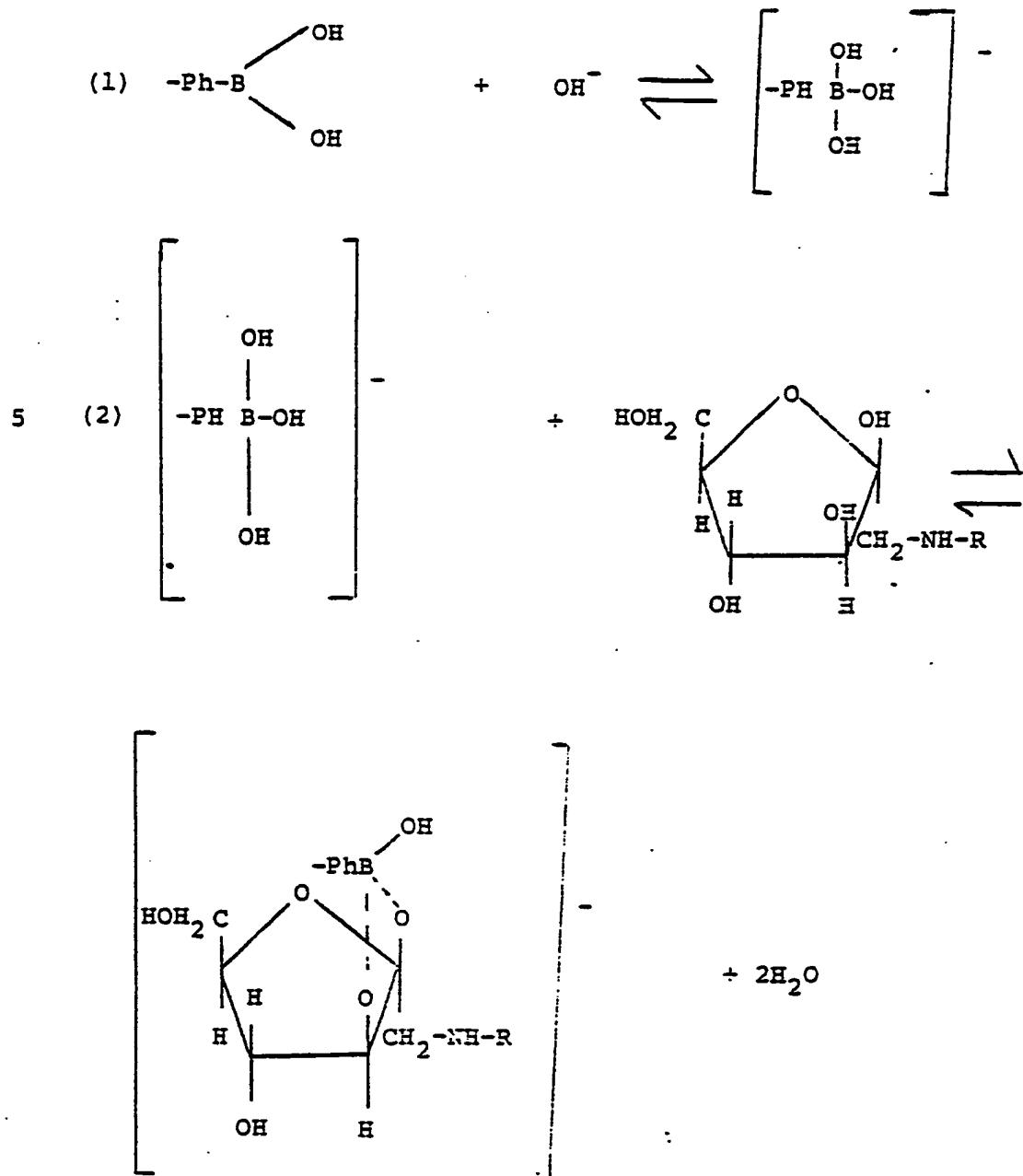
In contrast to the testing of urine for sugar level, which is indicative of glucose levels over only a few hours preceding the test; the technique of this invention, quantitation of the non-enzymatically glycosylated amino acids and peptides, provides a reflection of the integrated glucose concentration over the period of about one week to one month preceding the test. Accordingly, by separating and quantitating the non-enzymatically glycosylated amino acids and peptides found in urine, it is possible to obtain an accurate reflection of the integrated mean glucose concentration. Furthermore, this method is readily adaptable to many clinical environments since it involves only urine sampling. In particular, the method is not only readily useable in the hospital laboratory, but is also readily used for outpatient services and in the office of the treating physician.

The method is also based on the observation that boronic acids in alkaline solution form specific complexes with the coplanar cis-diol groups of fructose (Ferrier, R.J., "Carbohydrate Boronates", Adv. Carb. Chem. Biochem., 35, 31-80 (1978)). Since the stable forms of non-enzymatic glycosylation products are known to be 1-deoxyfructosyl-derivatives of  $\epsilon$  -amino groups of lysine and 2-amino groups of N-terminal amino acids (Koenig et al, "Structure Of Carbohydrate Of Hemoglobin A<sub>1C</sub>", J. Biol. Chem., 252, 2992-2997 (1977)), boronic acids are utilized to specifically retain glycosylated amino acids and/or peptides. Preferably, the boronic acids are immobilized on a support so as to effect complex formation and separation simultaneously. However, the free boronic acids may also be used for complex formation followed by separation by contacting the complex with a suitable support, e.g., O-(carboxymethyl) cellulose, which has been converted into the acyl azide, which will react with an amino-phenylboronic



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acid complex. The principle of the adsorption is illustrated by the following equations using a phenyl boronic acid:

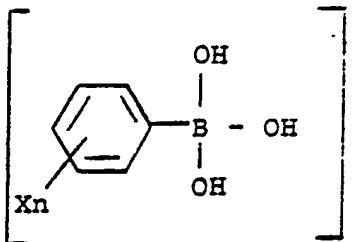


wherein Ph is a phenyl group and R is the remainder of the amino acid or peptide. Preferably, the p*z* is at

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least 9 during complex formation. Likewise, lowering the pH by treatment with an acid, e.g., HCl, will drive the equilibrium to the left thereby liberating the non-enzymatically glycosylated amino acid or peptide from the complex with the arylboronic acid. Quantitation may then be achieved by use of any conventional amino acid assay.

Suitable boronic acids include any boronic acid having two acid hydrogen atoms available for complex formation with the coplanar cis-diol groups of the non-enzymatically glycosylated amino acids and/or peptides. The cyclic anhydrides of such acids are also suitable, as are more complex boronates. A preferred group of the boronic acids are the arylboronic acids corresponding to the formula



wherein X is at least one member selected from the group consisting of amino, substituted amino, alkyl, halo and hydroxy and n is an integer of from 1 to 5.

In order to facilitate separation of the non-enzymatically glycosylated amino acids and/or peptides complexed with the boronic acids from the urine sample, the boronic acid is immobilized on a support either before or after complex formation. In other words, the boronic acid may be initially immobilized on a support and then contacted with the non-enzymatically glycosylated amino acids and/or peptides, thereby separating these amino acids and/or peptides from solution; or the boronic acid in free form may be contacted with the non-enzymatically glycosylated amino acids and/or pep-

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tides and then immobilized, thereby effecting separation of the complex.

Any suitable support, conventionally used for the immobilization of boronic acids, is useable in this 5 invention. Illustrative of such supports are plastics or polymers, e.g., polyacrylamide, cellulose or cellulose derivatives, e.g., paper, carboxymethylcellulose, aminoethylcellulose, etc., or glass, e.g., glass beads.

The amount of the boronic acid used is not critical, so long as sufficient material is provided to 10 complex with all the non-enzymatic glycosylated amino acids and/or peptides. This amount is readily determinable based on the sample aliquot size to be tested and a knowledge of the expected limits of concentration. 15 Typically, about 5-50 grams of a supported boronic acid (support + acid) are suitable for use with an aliquot (about 0.5%) of a 24 hour urine sample.

After separation of the complex (insolubilized boronic acid complexed with the non-enzymatically 20 glycosylated amino acids and/or peptides) by conventional liquid-solid separation techniques, e.g., filtration, the complex can then be directly analyzed for amino acid content; or the complex can be broken by acidification, followed by separation of the immobilized boronic acid, e.g., by filtration, and analysis 25 of the filtrate for amino acid content.

Many methods for amino acid quantitative analysis exist, e.g., the Millon reaction (tyrosine), Xanthoproteic reaction (tyrosin, tryptophan, phenylalanine), 30 Hopkins-Cole reaction (tryptophan), Ehrlich reaction (tryptophan), Sakaguchi reaction (arginine), Nitroprusside reaction (cysteine), Sullivan reaction (cysteine), Pauly reaction (histidine, tyrosine), Folin-Ciocalteu reaction (tyrosine), ninhydrin reaction ( $\alpha$ - 35 and  $\epsilon$ -amino groups), and these are all applicable to



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the present invention. While many of these tests exhibits specificity for certain amino acids the methods are completely applicable, since urine is a highly heterogeneous mixture of amino acids.

5 In view of simplicity of operation, color dye reactions, analogous to the Pauly reaction (diazotized sulfanilic acid), are especially effective in that they can be carried out directly on the complex. Azo dyes are especially suitable, e.g., "dabsyl chloride" (4-dimethylamino azobenzene 4'-sulfinyl chloride). Other useful color reactions may also be used, e.g., "dansyl chloride" (1-dimethylamino naphthalene-5-sulfonyl chloride) which detects amino groups, alkyl and aryl isothiocyanates, e.g., phenylisothiocyanate, various sulfonyl chlorides can covalently react with amino groups which can be quantitated.

Preferably, the color dye reaction is carried out prior to complex formation, since the attached chromophore will not interfere with complex formation, 20 attachment of the chromophore occurring at sites other than the non-enzymatically glycosylated sugar.

In general, Applicants' method comprises separating the non-enzymatically glycosylated amino acid and/or peptides from a urine sample and then quantitating the so-separated material. More particularly, 25 separation is achieved by selective complexation of the non-enzymatically glycosylated amino acids and/or peptides in the urine. Preferably, such complexation is achieved by treatment with immobilized boronic acids under alkaline pH conditions. The amount of the non-enzymatically glycosylated amino acids and/or peptides is then assayed by standard tests for amino acids content. The results of this assay which is reflective of 30 the integrated blood glucose concentration, can then be used by the clinician to determine appropriate modes of



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therapy. e.g., insulin, diet, exercise, etc., for the diabetic patient.

Additionally, the method of this invention and the test kit of this invention, described hereinafter, have  
5 been found to be equally applicable to the field of veterinary medicine, particularly, with respect to the treatment of household pets and the dietary management of food-producing animals.

The method of this invention is readily facilitated by the provision of a test kit comprising an insolubilized boronic acid and a color dye reactive with amino acids. In particular, the insolubilized boronic acid and the color dye are enclosed in separate packets within the kit. Any means of packaging may be used, so  
10 long as the two materials are kept separate until combination of the same is desired. Preferably, the insolubilized boronic acid is equilibrated with an alkaline buffer prior to packaging. Preferably, the color dye reactive with amino acids is an azo dye, e.g.,  
15 diazotized sulfanilic acid or 4-dimethylamino azobenzene 4-sulfinyl chloride, or "dansyl chloride" (1-dimethylamino naphthalene-5-sulfonyl chloride).

The preferred form of the test kit comprises a test strip or test stick and a dye packet. The test  
25 strip, illustrated in Figure 6, comprises a strip of porous material (1) upon which has been coated the boronic acid (2). The strip of porous material can be any material which is inert to the test conditions. Particularly preferred materials are those previously  
30 described as being capable of immobilizing boronic acids, thereby allowing direct immobilization of the boronic acid on the strip. Alternatively, an immobilized boronic acid may be coated on the strip.

Likewise, the test stick, illustrated in Figure 7,  
35 may be formed of the same porous materials (3) on which



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the same boronic acid materials (4) have been coated. While the figure illustrates a circular cross-section, any suitable cross-section can be used.

Figure 8 illustrates a dye packet for use in the 5 test kit of this invention, however, any suitable packaging means may be used. Typically an envelope (5) is used to contain the dye (6).

Sufficient dye is provided so as to be in at least stoichiometric excess of that necessary to react with 10 all of the amino acids present in the urine sample to be tested.

Sufficient boronic acid is provided so as to be in at least stoichiometric excess of that necessary to form a complex with the non-enzymatically glycosylated 15 amino acids and peptides present in the urine sample.

While there is some variation as to total amino acid content of a urine sample, those skilled in the art can readily determine the stoichiometric amount of dye to be used. Additionally, the non-enzymatically 20 glycosylated amino acid and/or peptide content is subject to variation, e.g., from about 24 to 48 or higher  $\mu$  moles leucine equivalent/24 hrs/kg of body wt., however, those skilled in the art can readily determine the stoichiometric amount of boronic acid to 25 be used..

The most preferred form of the test kit comprises a test strip which has been at least partially coated with a boronic acid and then impregnated with a color dye reactive with amino acids. Suitable amounts of the 30 component materials are as described above. The same materials and techniques for production of the test strip, described above, can be used with the additional step of impregnating the test strip with the dye. By this technique, a urine sample may be assayed by con-



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tacting a pH adjusted uring sample with the test strip, washing the test strip free of non-complexed compounds and non-reacted dye, and then colorimetrically comparing the strip with previously prepared standards.

5       The methodology of use of the test kit is quite simple. First a urine sample, previously adjusted to an alkaline pH, is reacted with the appropriate dye and then, second, the colored solution is brought into contact with the insolubilized boronic acid. third, the  
10      boronic acid is washed free of non-complexed compounds and non-reacted dye with an alkaline buffer. The remaining colored insolubilized boronic acid complex can then be colorimetrically compared with previously prepared standards to determine the non-enzymatically  
15      glycosylated amino acid and/or peptide concentration.

Having generally described this invention, a further understanding can be obtained by reference to certain specific examples which are provided herein for purposes of illustration only and are not intended to  
20      be limiting unless otherwise specified.

The following procedure was utilized, unless otherwise specified, in the following examples. Aliquots of samples (0.54% of a 24 hour urine sample) were adjusted to pH 9.0 by adding 0.1 M NaOH. After this pH  
25      adjustment the sample was shaken for 30 seconds on a vortex and then centrifuged (5 minutes, 12,000 x g) to remove insoluble material. The supernatant fluid was then transferred to the top of a 1.5 x 14 cm boronate column previously equilibrated with 0.025 M sodium  
30      phosphate buffer, pH 9.0, made 1.0 mM with NaN<sub>3</sub>. The sample was allowed to flow through the column until the liquid meniscus reached the top of the gel bed. Fresh buffer was then applied to the column and 4.0 ml fractions were collected at a flow rate of 20ml per hour.  
35      After 100 fractions were collected, the buffer was



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removed and the column was eluted with 0.025 M HCl. Fractions (2.0 ml) were collected at a flow rate of 20 ml per hour. Aliquots (400  $\mu$ ) from each fraction were assayed for the presence of ninhydrin-positive material, using the method of Moore and Stein ("A Modified Ninhydrin Reagent For The Photometric Determination Of Amino Acids And Related Compounds", J. Biol. Chem., 211, 907-913 (1954)). Each peak was integrated by weighing to determine the relative amount of ninhydrin-positive material present. The boronate column used comprises m-aminophenyl boronic acid immobilized on Bio-Gel P-6 (Affi-Gel 601, Bio-Rad Laboratories - Richmond, California).

Example 1

15 A synthetic mixture of  $^{14}\text{C}$ -glycosylated lysine (hydrolyzed N<sup>6</sup>-l-(deoxyglucitoyl) N<sup>2</sup>-t-butoxycarbonyl-lysine -- a gift from Dr. Victor Stevens) and  $^3\text{H}$ -lysine (New England Nuclear - Boston, Mass.) was prepared. When the synthetic mixture of radioactive standards was 20 analyzed, the vortex and centrifugation steps were omitted. Aliquots were counted in 7.0 ml hydrofluor (National Diagnostics - Sommerville, N.J.), in a Packard liquid scintillation counter. The results are shown in Figure 1, wherein synthetic  $^{14}\text{C}$ -glycosylated 25 lysine was specifically and completely separated from a mixture of the  $^{14}\text{C}$ -glycosylated lysine and unmodified  $^3\text{H}$ -lysine, demonstrating that breakdown products of non-enzymatically glycosylated proteins would be specifically adsorbed to the column.

30 Example 2

Twenty-four hour urine samples were collected from six normal volunteers and six unselected juvenile-onset diabetic patients. Toluene (10 ml) was added as a pre-



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servative during the collection period. Aliquots not analyzed immediately were frozen and stored at -70°C.

All urine samples gave essentially identical chromatographs. Washing the column with pH 9.0 sodium phosphate buffer produced a large peak of ninhydrin-positive material in the void column (Figure 2). After washing the column with 12 column volumes of buffer, changing to 0.025 M HCl resulted in the elution of a single peak of ninhydrin-positive material (Figure 3). Since the material in this latter peak must contain both a glycosylated amino group (to be retained on the column) and an unsubstituted amino group (to react with ninhydrin), it represents a mixture of free glycosylated lysine and small peptides containing glycosylated lysine.

Comparison of the level of these compounds in urine from diabetic and normal patients shows that the mean level in diabetics is two times that found in urine from normal subjects (Figure 4). These results show that the measurement of glycosylated proteins or peptides in the urine by this method provides a practical means of monitoring metabolic control in diabetic patients at frequent intervals at home.

Example 3

Human diabetic hemoglobin was prepared by the method of Koenig et al, (J. Biol. Chem., 252, 2992-2997 (1977)). Hemoglobin (25 mg) was reacted with tritiated sodium borohydride according to the procedure of Bookchin et al ("Structure Of Hemoglobin A<sub>1C</sub>: Nature Of The N-Terminal  $\beta$  -Chain Blocking Group", Biochem. Biophys. Res. Commun., 2, 86-93 (1968)), except that dialysis time was reduced to 24 hours. Globin samples were hydrolyzed in 6N HCl in sealed tubes at 105°C for 16 hours. After pH adjustment to 9.0, globin hydro-



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lysates were applied to the previously described boro-  
5 mate columns. Aliquots (200  $\mu$ l) from each fraction  
were counted in 7.0 ml of hydrofluor, in a Packard  
liquid scintillation counter. Peak tubes from the HCl  
10 column elution were pooled and concentrated. Samples  
were applied to a Beckman Model 119C amino acid ana-  
lyzer with stream division (90% to the fraction col-  
lector, 1.0 ml fractions) and aliquots (500  $\mu$ l) were  
counted in 7.0 ml of hydrofluor. The elution patterns  
15 were compared with chromatograms of reduced, hydrolyzed  
glycosylated valine and reduced, hydrolyzed glycosylated  
lysine standards. The chromatographic pattern of  
hydrolyzed diabetic hemoglobin previously reduced with  
15  $\text{NaBH}_4^3$  showed a peak of non-specific radioactivity in  
the void volume (non-glycosylated compounds) and a  
single sharp peak of radioactivity eluting subsequently  
with HCl. Amino acid analysis of this latter peak  
showed that reduced glycosylated valine and reduced  
glycosylated lysine in approximately equimolar amounts  
20 were the only adducts present in the pooled HCl peak.  
These findings are in agreement with previously pub-  
lished data obtained using other methodologies (Bunn et  
al, "Structural Heterogeneity Of Human Hemoglobin A Due  
To Non-Enzymatic Glycosylation", J. Biol. Chem., 254,  
25 3892-3989 (1979) and Gabbay et al, Glycosylated Hemo-  
globins: Increased Glycosylation Of Hemoglobin A In  
Diabetic Patients", Diabetes, 28, 337-340 (1979)).

Example 4

Using the procedure of Example 3, the peripheral  
30 (sciatic) nerve of normal and diabetic rats were ana-  
lyzed for the amount of non-enzymatic glycosylated  
amino acids. The nerves were removed, reduced with  
sodium borohydride, acid hydrolyzed and put on the  
boronate column as described previously. The results  
35 are shown in Figure 5, expressed as counts per minute



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(cpm) per micromole of ninhydrin positive material.

Having now fully described this invention, it will be apparent to one of ordinary skill in the art that many changes and modifications can be made thereto

5 without departing from the spirit or scope of the invention set forth herein.



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Claims

1. A method of separating and quantitating non-enzymatic glycosylated amino acids, peptides or mixtures thereof, comprising:

(a) treating a fluid containing said non-enzymatic glycosylated amino acids, peptides or mixtures thereof with a boronic acid to form a complex of said non-enzymatic glycosylated amino acids, peptides or mixtures thereof with said boronic acid;

(b) separating the so-formed complex from said fluid; and

(c) quantitatively analyzing the complexed material for amino acids, peptides or mixtures thereof.

2. A method of separating and quantitating non-enzymatic glycosylated amino acids, peptides or mixtures thereof, comprising:

(a) contacting a fluid containing said non-enzymatic glycosylated amino acids, peptides or mixtures thereof with a color dye reactive with amino acids so as to attach the dye to said non-enzymatic glycosylated amino acids, peptides or mixtures thereof;

(b) treating the dye containing solution so produced with a boronic acid to form a complex of said non-enzymatic glycosylated amino acids, peptides or mixtures thereof with such boronic acid;

(c) separating the so-formed complex from said fluid; and

(d) colorimetrically analyzing the dyed, complexed material for the quantity of amino acids, peptides or mixtures thereof.

3. The method according to Claim 1 or 2, wherein



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said complex is formed under alkaline conditions.

4. The method according to Claim 3, wherein said alkaline conditions comprise a pH of at least 9.

5. The method according to Claim 1 or 2, wherein said boronic acid is immobilized on a support.

6. The method according to Claim 1 or 2, wherein prior to analysis for amino acids, peptides or mixtures thereof the complex is treated with an acid.

7. The method according to Claim 6, wherein said acid is HCl.

8. The method according to Claim 1 or 2, wherein said fluid containing non-enzymatic glycosylated amino acids, peptides or mixtures thereof is a urine sample.

9. The method according to Claim 8, wherein naturally occurring insoluble materials have been removed from the urine sample.

10. A method for monitoring metabolic control in a diabetes patient comprising measuring the amount of non-enzymatic glycosylated amino acids and peptides in the urine of the patient.

11. The method according to Claim 9, wherein the integrated blood glucose concentration is monitored.

12. The method according to Claim 10, wherein the non-enzymatic glycosylated amino acids and peptides present in the urine of the patient are separated from the urine and quantitated.



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13. The method according to Claim 11, wherein the non-enzymatic glycosylated amino acids and peptides are separated from the urine by treatment with an insolubilized boronic acid so as to form an insoluble complex of said boronic acid and said non-enzymatic glycosylated amino acids and peptides; and separating said complex from the urine.

14. The method according to Claim 12, wherein prior to complex formation, the insoluble materials present in the urine have been removed.

15. A test kit for the determination of non-enzymatically glycosylated amino acids and/or peptides in a urine sample comprising an insolubilized boronic acid and a color dye reactive with amino acids, wherein said boronic acid and said color dye are separated from one another.

16. The test kit according to Claim 15, wherein the insolubilized boronic acid has been equilibrated with an alkaline buffer.

17. The test kit according to Claim 15, wherein the color dye is an azo dye.

18. The test kit according to Claim 15, wherein the color dye is dansyl chloride.

19. The test kit according to Claim 15, wherein said boronic acid and said color dye are in separate packets such that a pH adjusted urine sample may be contacted with the color dye, brought into contact with said boronic acid, and the boronic acid can be washed free of the urine sample non-reacted dye.



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20. The test kit according to Claim 15, wherein said boronic acid is coated on a strip of porous material.

21. The test kit according to Claim 15, wherein said boronic acid is coated on an elongated cylindrical body comprising a porous material.

22. The test kit according to Claim 15, wherein said strip of porous material is coated on only one side.

23. The test kit according to Claim 15 or 20, wherein the color dye reactive with amino acids is present in an amount which is stoichiometrically in excess of that necessary to react with all of the amino acids present in the urine sample.

24. The test kit according to Claim 23, wherein the insolubilized boronic acid is present in an amount which is stoichiometrically in excess of that necessary to form a complex with the non-enzymatic glycosylated amino acids and peptides present in the urine sample.

25. A test strip for determination of non-enzymatically glycosylated amino acids and/or peptides in a urine sample comprising a strip of porous material which has been at least partially coated with an insolubilized boronic acid and impregnated with a color dye reactive with amino acids.

26. The test strip according to Claim 25, wherein the color dye reactive with amino acids is present in an amount which is stoichiometrically in excess of that necessary to react with all of the amino acids present in the urine sample.

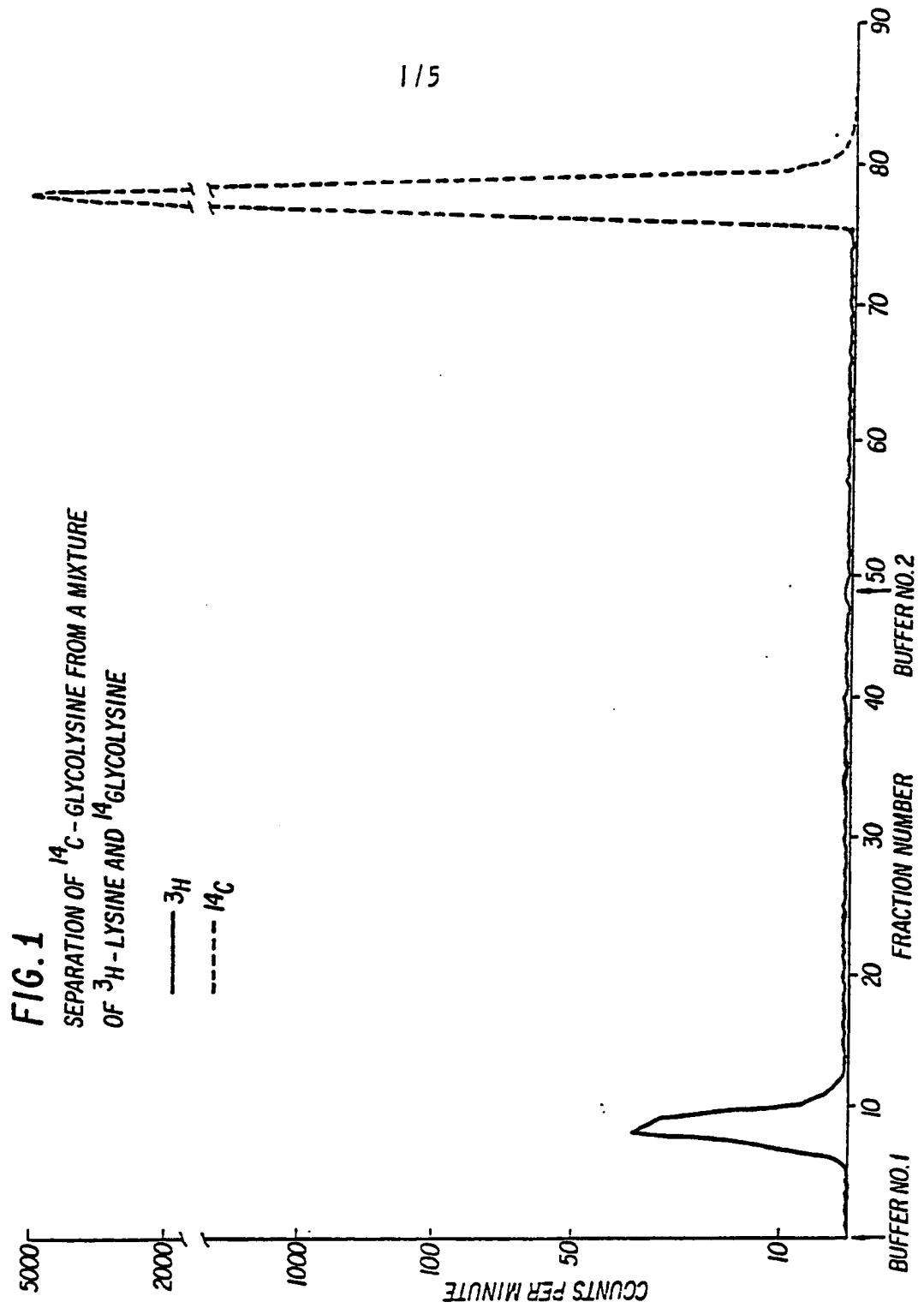


-21-

27. The test strip according to Claim 26, wherein the insolubilized boronic acid is present in an amount which is stoichiometrically in excess of that necessary to form a complex with the non-enzymatic glycosylated amino acids and peptides present in the urine sample.

28. The test strip according to Claim 27, wherein the strip of porous material is paper.





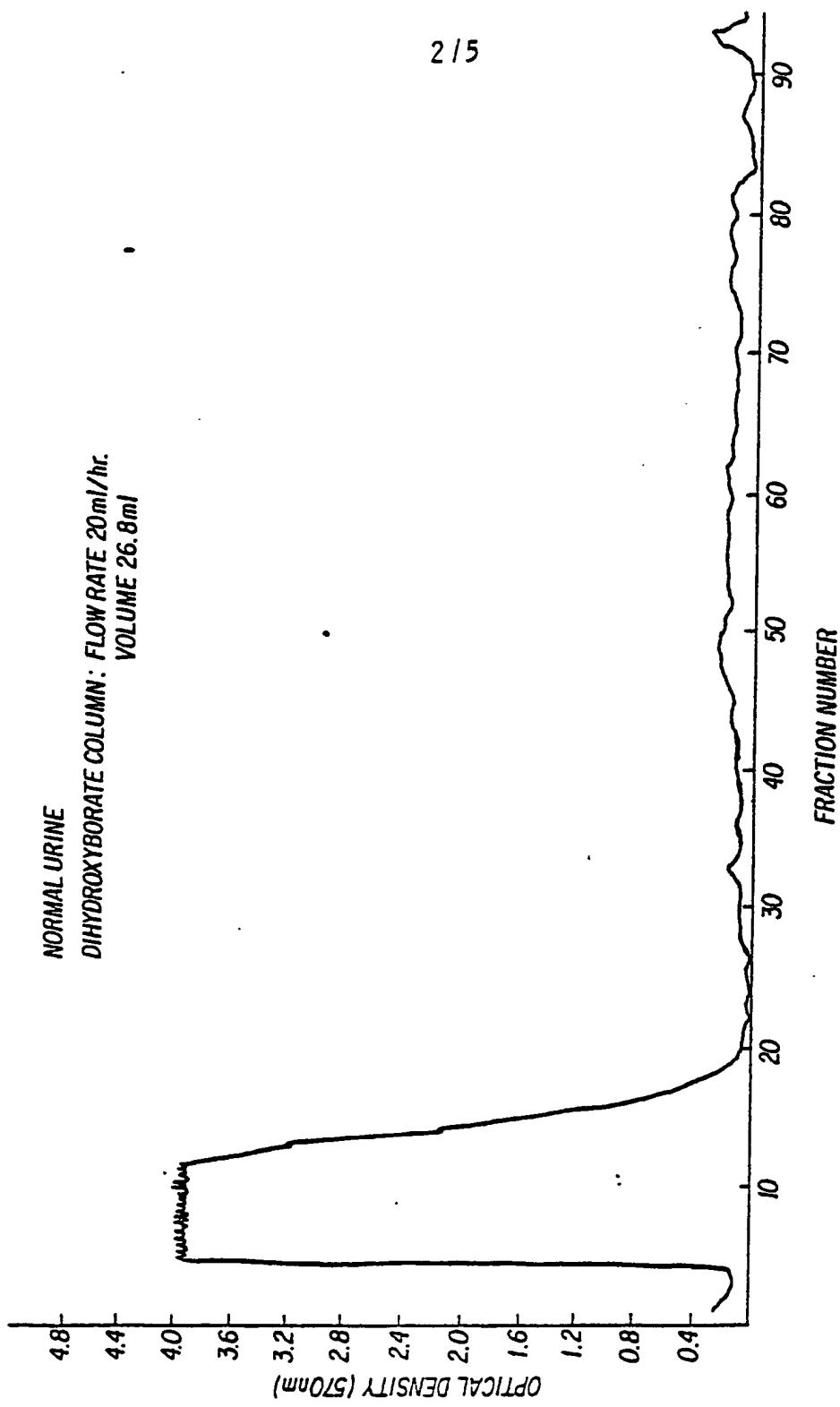
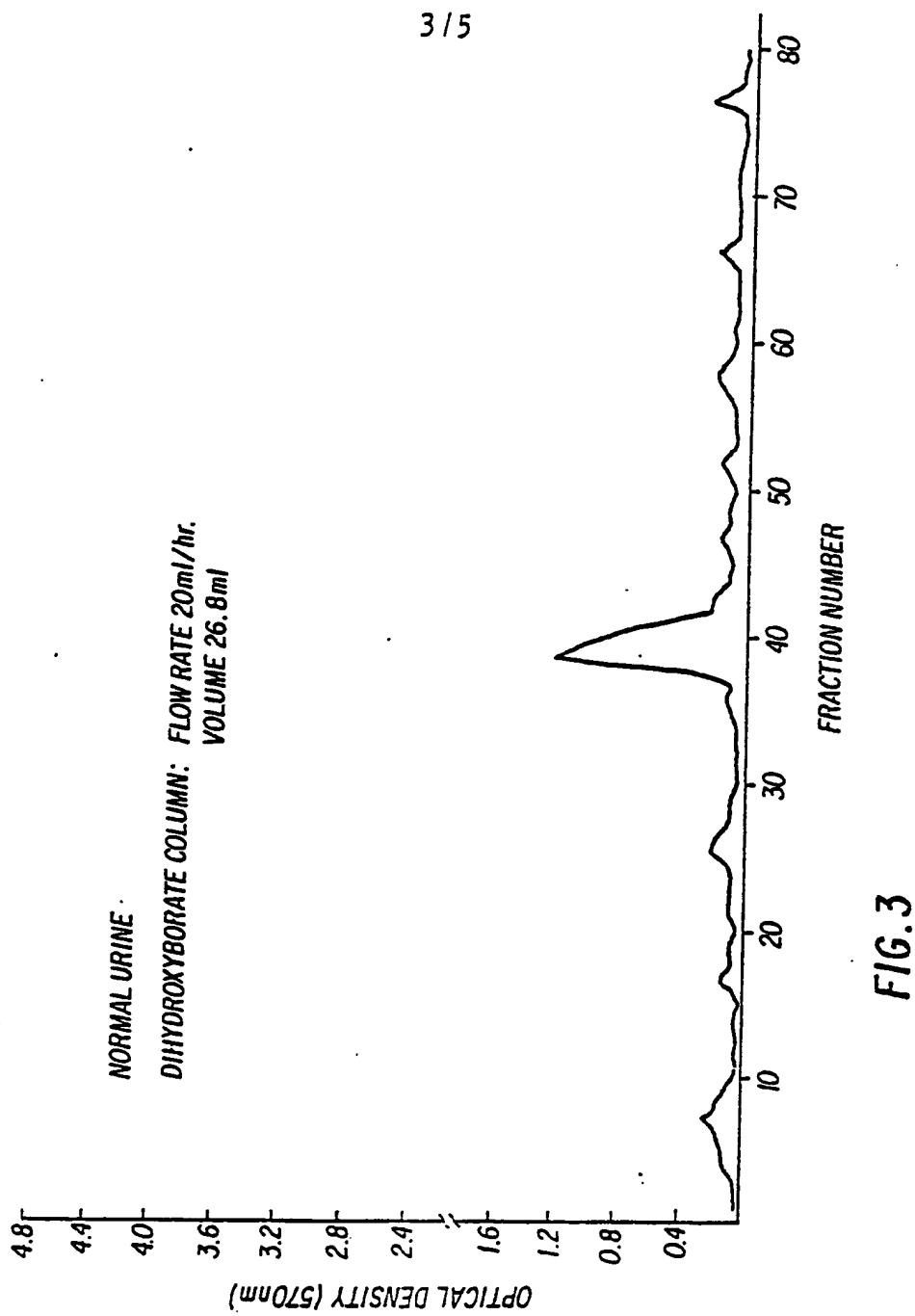


FIG. 2





SUBSTITUTE SHEET



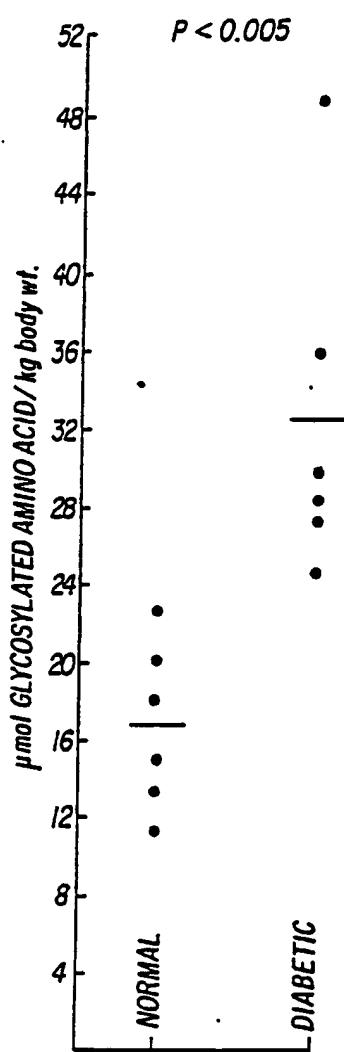


FIG. 4

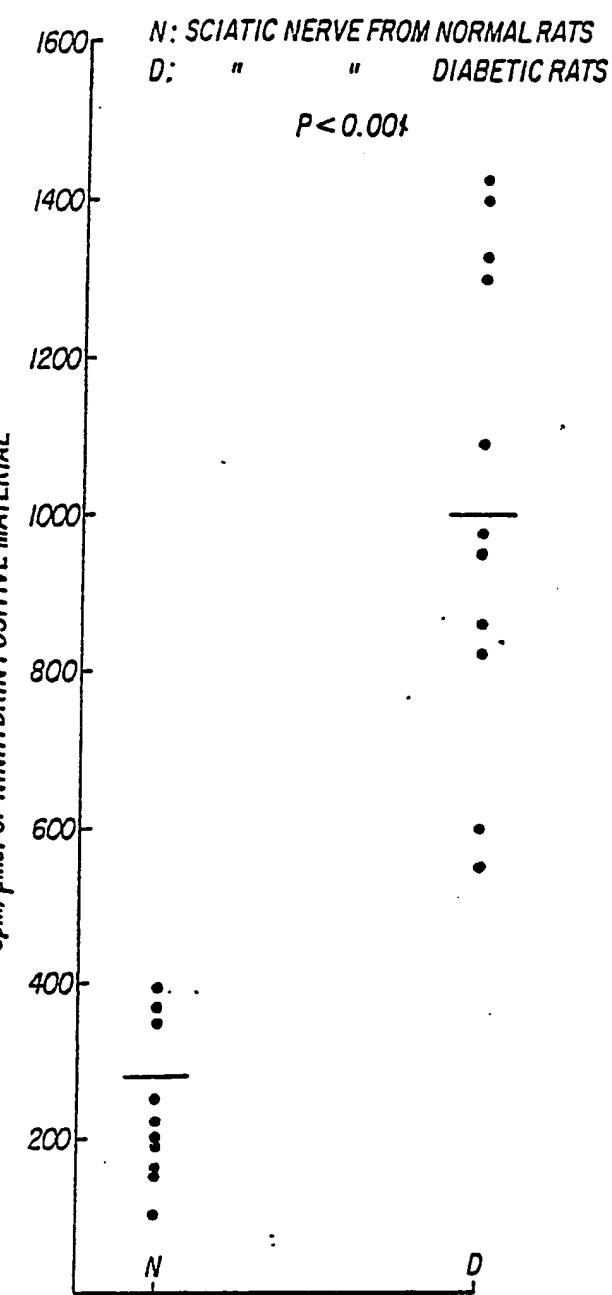


FIG. 5

SUBSTITUTE SHEET

BAD ORIGINAL



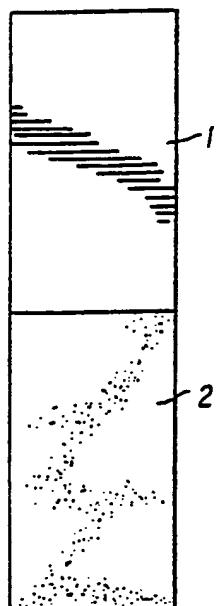


FIG. 6a

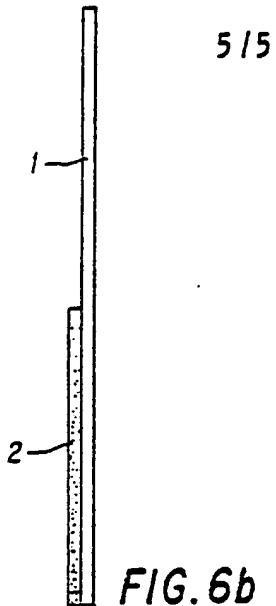


FIG. 6b

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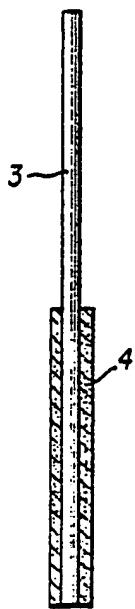


FIG. 7a

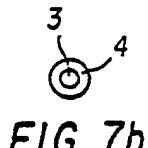


FIG. 7b

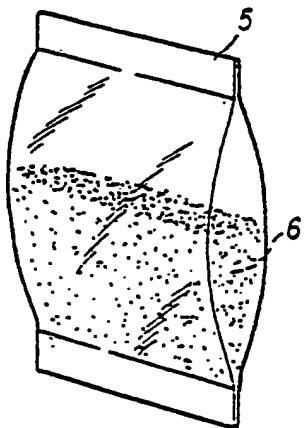
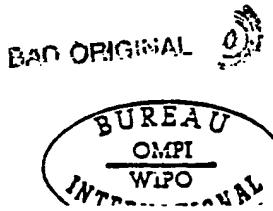


FIG. 8



# INTERNATIONAL SEARCH REPORT

International Application No PCT/US81/01525

## I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) \*

According to International Patent Classification (IPC) or to both National Classification and IPC  
 INT. CL 3 G01N 33/68  
 U.S. CL. 23/230B

## II. FIELDS SEARCHED

Minimum Documentation Searched *	
Classification System	Classification Symbols
U.S.	23/230B, 901, 913; 422/56, 57, 61
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched *	

## III. DOCUMENTS CONSIDERED TO BE RELEVANT \*\*

Category *	Citation of Document, ** with indication, where appropriate, of the relevant passages ***	Relevant to Claim No. ****
A	US,A, 4,269,605, PUBLISHED 26 MAY 1981, DEAN ET AL.	I - 28

\* Special categories of cited documents: \*\*

- "A" document defining the general state of the art
- "E" earlier document but published on or after the international filing date
- "L" document cited for special reason other than those referred to in the other categories
- "O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but on or after the priority date claimed

"T" later document published on or after the international filing date or priority date and not in conflict with the application, but cited to understand the principle or theory underlying the invention

"X" document of particular relevance

## IV. CERTIFICATION

Date of the Actual Completion of the International Search :

02 FEBRUARY 1982

Date of Mailing of this International Search Report :

19 FEB 1982

International Searching Authority :

ISA/US

Signature of Authorized Officer \*\*

PETER CHEN